

N-(4-Methylpyridin-2-yl)-5H-dibenzo-*[d,f]*[1,3]diazepine-6-carboxamide toluene hemisolvate

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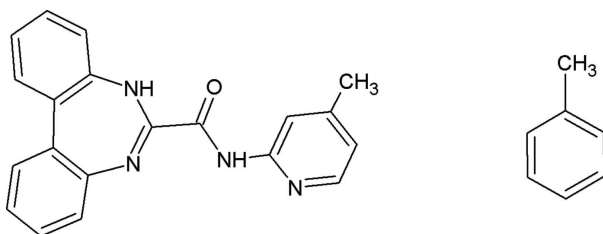
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; disorder in solvent or counterion; R factor = 0.052; wR factor = 0.148; data-to-parameter ratio = 30.0.

The title compound, $\text{C}_{20}\text{H}_{16}\text{N}_4\text{O} \cdot 0.5\text{C}_7\text{H}_8$, crystallizes with half a toluene molecule (disordered about a centre of inversion); the seven-membered diazepine ring adopts a twist-boat conformation. The structure features two intramolecular ($\text{N}-\text{H} \cdots \text{O}$ and $\text{N}-\text{H} \cdots \text{N}$) and one intermolecular hydrogen bond. Additional intermolecular $\pi-\pi$ interactions (3.556–3.651 Å) give rise to a three-dimensional network in the crystal structure.

Related literature

For literature on the medicinal and psychotherapeutic uses of similar compounds, see Evans *et al.* (1988); Sternbach (1978). For comparative bond distances, see Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{16}\text{N}_4\text{O} \cdot 0.5\text{C}_7\text{H}_8$ $a = 8.9323$ (2) Å
 $M_r = 374.44$ $b = 9.6863$ (2) Å
 Triclinic, $P\bar{1}$ $c = 12.1478$ (3) Å

$\alpha = 77.050$ (1)°
 $\beta = 87.529$ (1)°
 $\gamma = 70.191$ (1)°
 $V = 963.08$ (4) Å³
 $Z = 2$

Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 293$ (2) K
 $0.29 \times 0.21 \times 0.15$ mm

Data collection

Nonius KappaCCD diffractometer
 Absorption correction: multi-scan
 (*DENZO* and *SCALEPACK*;
 Otwinowski & Minor, 1997)
 $T_{\min} = 0.977$, $T_{\max} = 0.988$
 17061 measured reflections
 8631 independent reflections
 5192 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.0595$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.148$
 $S = 1.02$
 8631 reflections
 288 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{N3}-\text{H3} \cdots \text{O1}$	0.88 (1)	2.22 (1)	2.677 (1)	112 (1)
$\text{N3}-\text{H3} \cdots \text{O1}^i$	0.88 (1)	2.31 (1)	3.108 (1)	151 (1)
$\text{N2}-\text{H2} \cdots \text{N4}$	0.87 (1)	2.19 (1)	2.644 (1)	112 (1)

Symmetry code: (i) $-x, -y + 1, -z + 1$.

Data collection: *COLLECT* (Nonius, 2000); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *SCALEPACK* and *DENZO* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2316).

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supplementary materials

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***N*-(4-Methylpyridin-2-yl)-5*H*-dibenzo[*d,f*][1,3]diazepine-6-carboxamide toluene hemisolvate**

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Comment

Benzodiazepines are one of the most important substances in medicinal chemistry, representing the prototypical 'privileged structure' (Evans *et al.* 1988). Compounds based on the diazepine skeleton are psychotherapeutic agents widely used for the treatment of anxiety and neurosis (Sternbach, 1978). The crystal structure analysis of *N*-(4-methylpyridin-2-yl)-5*H*-dibenzo[*d,f*][1,3] diazepine-6-carboxamide was performed to find out the influence of methyl substituent on molecular geometry and packing properties. The molecular structure of the title compound is shown in Fig. 1. The symmetrically independent part of the unit cell is composed of title compound molecule and disordered toluene molecule. The core of the diazepine molecule is composed of a seven-membered ring to which a pyridin-2-ylamino carbonyl substituent is attached and two benzene rings are fused. The puckering parameters of the seven-membered ring (atoms in N3, C2, ..., C31 sequence): $q_2 = 0.5752$, $q_3 = 0.0819$, $QT = 0.5810$, $\varphi_2 = -6.33$, $\varphi_3 = 10.46$, $\theta_2 = 81.89^\circ$, indicate a twisted-boat conformation with a pseudo-mirror plane (C_s) through the N3 atom and the centre of C41—C46 bond. From the chemical point of view there is also a pseudo 2-fold axis (C_2) bisecting the diazepine ring through C2 and the center of C36—C46 bond. The bond lengths and valence angles are typical and comparable with the values reported in the literature (Allen *et al.* 1987). Two N atoms are involved in hydrogen bonds: two intra- and one intermolecular (Table 1). The intermolecular N3—H3...O1 [$-x, 1 - y, 1 - z$] interaction is relatively weak due to simultaneous donor participation in intramolecular interaction (bifurcated hydrogen bond). In addition, there are two types of π - π interactions, which assist in the stabilization of the three-dimensional-packing. The benzene ring C31...C36 is strictly parallel to another one in position [$1 - x, 1 - y, 1 - z$], with interplanar spacing of 3.556 Å, the ring-centroid separation of 3.883 Å and with ring offset of 1.560 Å. The pyridine ring is nearly parallel to another benzene ring C41...C46 of the molecule at [$1 - x, -y, 1 - z$]. The interplanar spacing is 3.651 Å, the corresponding ring-centroid separation is 3.790 Å, resulting in the offset of 1.017 Å. The π - π interaction are illustrated in Fig. 2. As it can be seen in Fig. 2, the solvent accessible cavities at 0, 1/2, 1 are filled with disordered toluene molecules which stabilized three-dimensional structure (total potential solvent area of 195.7 Å³ makes 20.3% of unit cell volume).

Experimental

Recrystallization from toluene afforded crystals suitable for X-ray measurements.

Refinement

H atoms bonded to N atoms were located in a difference Fourier map and refined with distance restraints of N—H = 0.87 (2) Å, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$. All other H atom positions were also observed in difference Fourier map. Nevertheless, in the refinement procedure the hydrogen atoms were positioned geometrically and refined using a riding model (including free rotation about the C—C bond), with C—H = 0.95–0.99 Å (C—H = 0.97 Å for CH₂ groups, 0.96 Å for CH₃ groups, and 0.93 Å for aromatic CH) and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl groups and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for all other H atoms. Refinement was performed with data set combined of two independent measurements and with BASF parameter as the scale factor for second data set (BASF parameter converged to 0.94477)

Figures

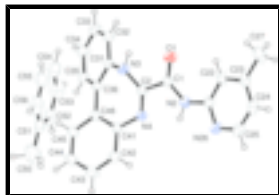


Fig. 1. ORTEP-3 (Farrugia, 1997) drawing of the title compound with labels. Displacement ellipsoids of non-H atoms drawn at 30% probability level.

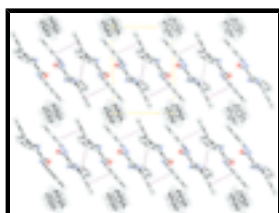


Fig. 2. ORTEP-3 (Farrugia, 1997) drawing of the crystal packing viewed along [010] with marked π - π interactions (dashed bonds). Hydrogen atoms are omitted.

N-(4-Methylpyridin-2-yl)-5*H*-dibenzo[*d,f*][1,3]diazepine-6- carboxamide toluene hemisolvate

Crystal data

$C_{20}H_{16}N_4O \cdot 0.5C_7H_8$

$M_r = 374.44$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 8.9323$ (2) Å

$b = 9.6863$ (2) Å

$c = 12.1478$ (3) Å

$\alpha = 77.050$ (1)°

$\beta = 87.529$ (1)°

$\gamma = 70.191$ (1)°

$V = 963.08$ (4) Å³

$Z = 2$

$F_{000} = 394$

$D_x = 1.291$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 2469 reflections

$\theta = 1.0$ – 27.5 °

$\mu = 0.08$ mm⁻¹

$T = 293$ (2) K

Prism, orange

$0.29 \times 0.21 \times 0.15$ mm

Data collection

Nonius KappaCCD
diffractometer

8631 independent reflections

Radiation source: fine-focus sealed tube

5192 reflections with $I > 2\sigma(I)$

Monochromator: horizontally mounted graphite crystal

$R_{int} = 0.0000$

Detector resolution: 9 pixels mm⁻¹

$\theta_{max} = 27.4$ °

$T = 293$ (2) K

$\theta_{min} = 2.4$ °

ϕ and ω scans to fill asymmetric unit

$h = -11 \rightarrow 11$

Absorption correction: multi-scan

(DENZO and SCALEPACK; Otwinowski & Minor, 1997)

$k = -12 \rightarrow 12$

$T_{min} = 0.977$, $T_{max} = 0.988$

$l = 0 \rightarrow 15$

17061 measured reflections

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.052$	$w = 1/[\sigma^2(F_o^2) + (0.0656P)^2 + 0.0957P]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.148$	$(\Delta/\sigma)_{\max} = 0.01$
$S = 1.02$	$\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$
8631 reflections	$\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$
288 parameters	Extinction correction: SHELXL97 (Sheldrick, 1997), $F_c^* = kF_c[1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.014 (3)
Secondary atom site location: difference Fourier map	

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement with redundant (*i.e.* not merged/unique) data set: data set combined of two independent measurements; refinement with BASF parameter as the scale factor for second data set. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	0.17801 (13)	0.23113 (12)	0.51837 (9)	0.0458 (3)	
O1	0.07918 (10)	0.32376 (9)	0.56173 (8)	0.0611 (2)	
C2	0.27718 (12)	0.28173 (12)	0.42283 (9)	0.0430 (3)	
N3	0.23275 (11)	0.43460 (10)	0.38922 (9)	0.0505 (3)	
H3	0.1546 (16)	0.4821 (14)	0.4286 (10)	0.058 (4)*	
C31	0.34450 (12)	0.51109 (11)	0.36175 (9)	0.0435 (3)	
C32	0.32397 (14)	0.63436 (12)	0.40814 (10)	0.0534 (3)	
H32	0.2384	0.6645	0.4540	0.064*	
C33	0.42946 (16)	0.71276 (13)	0.38688 (11)	0.0605 (3)	
H33	0.4140	0.7963	0.4174	0.073*	
C34	0.55721 (15)	0.66709 (14)	0.32055 (12)	0.0603 (3)	
H34	0.6297	0.7184	0.3071	0.072*	
C35	0.57746 (14)	0.54485 (13)	0.27409 (10)	0.0521 (3)	

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H35	0.6645	0.5149	0.2294	0.062*	
C36	0.47134 (12)	0.46415 (11)	0.29185 (9)	0.0427 (2)	
C46	0.49668 (12)	0.33809 (12)	0.23420 (9)	0.0430 (2)	
C45	0.56872 (14)	0.34457 (14)	0.13001 (10)	0.0560 (3)	
H45	0.5935	0.4302	0.0970	0.067*	
C44	0.60463 (17)	0.22908 (16)	0.07395 (12)	0.0687 (4)	
H44	0.6513	0.2380	0.0040	0.082*	
C43	0.57127 (17)	0.10084 (17)	0.12164 (12)	0.0702 (4)	
H43	0.5972	0.0214	0.0852	0.084*	
C42	0.49880 (14)	0.09106 (14)	0.22422 (11)	0.0565 (3)	
H42	0.4765	0.0039	0.2565	0.068*	
C41	0.45811 (12)	0.20788 (12)	0.28090 (9)	0.0428 (2)	
N4	0.38063 (10)	0.17891 (10)	0.38303 (8)	0.0450 (2)	
N2	0.20866 (12)	0.08255 (11)	0.54501 (9)	0.0513 (3)	
H2	0.2779 (16)	0.0327 (15)	0.5026 (12)	0.069 (4)*	
C21	0.12759 (13)	-0.00035 (12)	0.61882 (9)	0.0458 (3)	
C22	0.03000 (14)	0.05641 (14)	0.70102 (10)	0.0531 (3)	
H22	0.0155	0.1528	0.7103	0.064*	
C23	-0.04599 (15)	-0.03245 (15)	0.76942 (10)	0.0577 (3)	
C24	-0.01902 (16)	-0.17409 (15)	0.75086 (11)	0.0645 (4)	
H24	-0.0679	-0.2374	0.7946	0.077*	
C25	0.08051 (16)	-0.22065 (14)	0.66741 (12)	0.0624 (3)	
H25	0.0970	-0.3166	0.6565	0.075*	
N26	0.15561 (12)	-0.13681 (10)	0.60055 (8)	0.0539 (3)	
C27	-0.1531 (2)	0.0259 (2)	0.86000 (13)	0.0853 (5)	
H27A	-0.1950	-0.0498	0.8997	0.102*	
H27B	-0.2393	0.1146	0.8260	0.102*	
H27C	-0.0933	0.0502	0.9121	0.102*	
C50	0.0601 (10)	0.3448 (10)	-0.1225 (8)	0.141 (4)	0.50
H50A	0.0559	0.4006	-0.1991	0.169*	0.50
H50B	-0.0135	0.2908	-0.1153	0.169*	0.50
H50C	0.1659	0.2749	-0.1032	0.169*	0.50
C51	0.0177 (5)	0.4482 (4)	-0.0469 (2)	0.0800 (9)	0.50
C52	0.1356 (3)	0.4569 (3)	0.0196 (3)	0.0699 (17)	0.50
H52	0.2411	0.3975	0.0150	0.084*	0.50
C53	0.0958 (5)	0.5544 (4)	0.0928 (2)	0.0986 (13)	0.50
H53	0.1747	0.5603	0.1373	0.118*	0.50
C54	-0.0619 (6)	0.6433 (4)	0.0996 (3)	0.118 (4)	0.50
H54	-0.0886	0.7085	0.1486	0.142*	0.50
C55	-0.1799 (3)	0.6345 (4)	0.0332 (4)	0.1057 (14)	0.50
H55	-0.2854	0.6940	0.0377	0.127*	0.50
C56	-0.1400 (4)	0.5370 (5)	-0.0401 (3)	0.099 (3)	0.50
H56	-0.2189	0.5312	-0.0845	0.119*	0.50

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0459 (6)	0.0426 (6)	0.0505 (6)	-0.0172 (5)	0.0101 (5)	-0.0112 (5)

O1	0.0620 (5)	0.0468 (5)	0.0748 (6)	-0.0191 (4)	0.0306 (5)	-0.0182 (4)
C2	0.0421 (5)	0.0398 (6)	0.0486 (6)	-0.0160 (4)	0.0072 (5)	-0.0103 (5)
N3	0.0454 (5)	0.0385 (5)	0.0656 (6)	-0.0132 (4)	0.0204 (5)	-0.0124 (4)
C31	0.0451 (6)	0.0366 (5)	0.0464 (6)	-0.0134 (4)	0.0077 (5)	-0.0062 (4)
C32	0.0592 (7)	0.0416 (6)	0.0570 (7)	-0.0142 (5)	0.0147 (6)	-0.0128 (5)
C33	0.0734 (8)	0.0438 (7)	0.0698 (8)	-0.0238 (6)	0.0090 (7)	-0.0187 (6)
C34	0.0628 (7)	0.0495 (7)	0.0760 (9)	-0.0293 (6)	0.0082 (7)	-0.0139 (6)
C35	0.0500 (6)	0.0488 (7)	0.0588 (7)	-0.0208 (5)	0.0128 (5)	-0.0101 (5)
C36	0.0435 (5)	0.0381 (6)	0.0446 (6)	-0.0137 (4)	0.0060 (5)	-0.0063 (4)
C46	0.0387 (5)	0.0444 (6)	0.0449 (6)	-0.0130 (4)	0.0064 (5)	-0.0104 (5)
C45	0.0584 (7)	0.0615 (7)	0.0519 (7)	-0.0253 (6)	0.0156 (6)	-0.0150 (6)
C44	0.0777 (9)	0.0797 (10)	0.0563 (8)	-0.0304 (7)	0.0272 (7)	-0.0292 (7)
C43	0.0783 (9)	0.0711 (9)	0.0721 (9)	-0.0259 (7)	0.0227 (8)	-0.0404 (7)
C42	0.0594 (7)	0.0505 (7)	0.0644 (8)	-0.0199 (6)	0.0138 (6)	-0.0221 (6)
C41	0.0389 (5)	0.0421 (6)	0.0459 (6)	-0.0114 (4)	0.0058 (5)	-0.0112 (4)
N4	0.0454 (5)	0.0399 (5)	0.0496 (5)	-0.0148 (4)	0.0107 (4)	-0.0110 (4)
N2	0.0558 (6)	0.0414 (5)	0.0557 (6)	-0.0171 (4)	0.0203 (5)	-0.0110 (4)
C21	0.0452 (6)	0.0430 (6)	0.0466 (6)	-0.0160 (5)	0.0030 (5)	-0.0033 (5)
C22	0.0604 (7)	0.0534 (7)	0.0495 (6)	-0.0248 (6)	0.0104 (6)	-0.0121 (5)
C23	0.0575 (7)	0.0733 (9)	0.0446 (6)	-0.0305 (6)	0.0065 (6)	-0.0053 (6)
C24	0.0656 (8)	0.0658 (9)	0.0628 (8)	-0.0354 (7)	0.0029 (7)	0.0049 (6)
C25	0.0651 (8)	0.0464 (7)	0.0745 (9)	-0.0238 (6)	0.0006 (7)	-0.0026 (6)
N26	0.0550 (6)	0.0419 (5)	0.0615 (6)	-0.0163 (4)	0.0049 (5)	-0.0058 (4)
C27	0.0911 (11)	0.1162 (13)	0.0609 (9)	-0.0534 (10)	0.0294 (8)	-0.0204 (9)
C50	0.189 (11)	0.164 (8)	0.103 (4)	-0.106 (7)	0.024 (5)	-0.027 (5)
C51	0.104 (3)	0.076 (2)	0.0657 (19)	-0.055 (2)	0.010 (2)	0.0105 (16)
C52	0.049 (3)	0.080 (4)	0.076 (2)	-0.037 (3)	-0.010 (2)	0.018 (2)
C53	0.100 (3)	0.114 (4)	0.089 (3)	-0.062 (3)	-0.022 (2)	0.008 (2)
C54	0.157 (9)	0.081 (4)	0.114 (6)	-0.049 (5)	0.037 (5)	-0.010 (4)
C55	0.081 (3)	0.105 (3)	0.117 (4)	-0.037 (2)	0.014 (3)	0.011 (3)
C56	0.096 (6)	0.114 (7)	0.084 (3)	-0.054 (5)	-0.007 (3)	0.015 (3)

Geometric parameters (Å, °)

C1—O1	1.223 (1)	N2—H2	0.87 (1)
C1—N2	1.335 (1)	C21—N26	1.327 (1)
C1—C2	1.524 (2)	C21—C22	1.380 (2)
C2—N4	1.278 (1)	C22—C23	1.385 (2)
C2—N3	1.364 (1)	C22—H22	0.9300
N3—C31	1.421 (1)	C23—C24	1.379 (2)
N3—H3	0.88 (1)	C23—C27	1.506 (2)
C31—C32	1.386 (2)	C24—C25	1.370 (2)
C31—C36	1.397 (1)	C24—H24	0.9300
C32—C33	1.381 (2)	C25—N26	1.340 (2)
C32—H32	0.9300	C25—H25	0.9300
C33—C34	1.373 (2)	C27—H27A	0.9600
C33—H33	0.9300	C27—H27B	0.9600
C34—C35	1.378 (2)	C27—H27C	0.9600
C34—H34	0.9300	C50—C51	1.455 (8)

supplementary materials

C35—C36	1.402 (2)	C50—H50A	0.9600
C35—H35	0.9300	C50—H50B	0.9600
C36—C46	1.488 (2)	C50—H50C	0.9600
C46—C45	1.393 (2)	C51—C52	1.3900
C46—C41	1.410 (2)	C51—C56	1.3900
C45—C44	1.377 (2)	C52—C53	1.3900
C45—H45	0.9300	C52—H52	0.9300
C44—C43	1.372 (2)	C53—C54	1.3900
C44—H44	0.9300	C53—H53	0.9300
C43—C42	1.378 (2)	C54—C55	1.3900
C43—H43	0.9300	C54—H54	0.9300
C42—C41	1.391 (1)	C55—C56	1.3900
C42—H42	0.9300	C55—H55	0.9300
C41—N4	1.414 (1)	C56—H56	0.9300
N2—C21	1.410 (1)		
O1—C1—N2	126.0 (1)	C2—N4—C41	123.5 (1)
O1—C1—C2	120.4 (1)	C1—N2—C21	128.8 (1)
N2—C1—C2	113.7 (1)	C1—N2—H2	114.6 (9)
N4—C2—N3	130.1 (1)	C21—N2—H2	115.8 (9)
N4—C2—C1	117.1 (1)	N26—C21—C22	124.3 (1)
N3—C2—C1	112.7 (1)	N26—C21—N2	112.9 (1)
C2—N3—C31	122.7 (1)	C22—C21—N2	122.8 (1)
C2—N3—H3	113.0 (8)	C21—C22—C23	118.9 (1)
C31—N3—H3	111.5 (8)	C21—C22—H22	120.5
C32—C31—C36	120.8 (1)	C23—C22—H22	120.5
C32—C31—N3	117.3 (1)	C24—C23—C22	117.4 (1)
C36—C31—N3	121.9 (1)	C24—C23—C27	122.3 (1)
C33—C32—C31	120.6 (1)	C22—C23—C27	120.3 (1)
C33—C32—H32	119.7	C25—C24—C23	119.4 (1)
C31—C32—H32	119.7	C25—C24—H24	120.3
C34—C33—C32	119.8 (1)	C23—C24—H24	120.3
C34—C33—H33	120.1	N26—C25—C24	124.2 (1)
C32—C33—H33	120.1	N26—C25—H25	117.9
C33—C34—C35	119.6 (1)	C24—C25—H25	117.9
C33—C34—H34	120.2	C21—N26—C25	115.7 (1)
C35—C34—H34	120.2	C23—C27—H27A	109.5
C34—C35—C36	122.3 (1)	C23—C27—H27B	109.5
C34—C35—H35	118.9	H27A—C27—H27B	109.5
C36—C35—H35	118.9	C23—C27—H27C	109.5
C31—C36—C35	116.8 (1)	H27A—C27—H27C	109.5
C31—C36—C46	124.0 (1)	H27B—C27—H27C	109.5
C35—C36—C46	119.1 (1)	C52—C51—C56	120.0
C45—C46—C41	117.3 (1)	C52—C51—C50	119.7 (5)
C45—C46—C36	118.8 (1)	C56—C51—C50	120.3 (5)
C41—C46—C36	123.9 (1)	C51—C52—C53	120.0
C44—C45—C46	122.5 (1)	C51—C52—H52	120.0
C44—C45—H45	118.7	C53—C52—H52	120.0
C46—C45—H45	118.7	C52—C53—C54	120.0
C43—C44—C45	119.8 (1)	C52—C53—H53	120.0

C43—C44—H44	120.1	C54—C53—H53	120.0
C45—C44—H44	120.1	C55—C54—C53	120.0
C44—C43—C42	119.2 (1)	C55—C54—H54	120.0
C44—C43—H43	120.4	C53—C54—H54	120.0
C42—C43—H43	120.4	C56—C55—C54	120.0
C43—C42—C41	121.9 (1)	C56—C55—H55	120.0
C43—C42—H42	119.1	C54—C55—H55	120.0
C41—C42—H42	119.1	C55—C56—C51	120.0
C42—C41—C46	119.2 (1)	C55—C56—H56	120.0
C42—C41—N4	113.7 (1)	C51—C56—H56	120.0
C46—C41—N4	127.1 (1)		
O1—C1—C2—N4	178.5 (1)	C45—C46—C41—C42	2.6 (2)
N2—C1—C2—N4	-2.8 (2)	C36—C46—C41—C42	-174.7 (1)
O1—C1—C2—N3	-4.9 (2)	C45—C46—C41—N4	-177.9 (1)
N2—C1—C2—N3	173.7 (1)	C36—C46—C41—N4	4.9 (2)
N4—C2—N3—C31	-44.4 (2)	N3—C2—N4—C41	-9.1 (2)
C1—C2—N3—C31	139.6 (1)	C1—C2—N4—C41	166.7 (1)
C2—N3—C31—C32	-132.8 (1)	C42—C41—N4—C2	-149.2 (1)
C2—N3—C31—C36	46.6 (2)	C46—C41—N4—C2	31.2 (2)
C36—C31—C32—C33	-0.5 (2)	O1—C1—N2—C21	6.7 (2)
N3—C31—C32—C33	178.9 (1)	C2—C1—N2—C21	-171.9 (1)
C31—C32—C33—C34	-0.9 (2)	C1—N2—C21—N26	160.4 (1)
C32—C33—C34—C35	1.1 (2)	C1—N2—C21—C22	-19.7 (2)
C33—C34—C35—C36	0.1 (2)	N26—C21—C22—C23	-0.5 (2)
C32—C31—C36—C35	1.7 (2)	N2—C21—C22—C23	179.6 (1)
N3—C31—C36—C35	-177.7 (1)	C21—C22—C23—C24	0.0 (2)
C32—C31—C36—C46	-176.7 (1)	C21—C22—C23—C27	179.9 (1)
N3—C31—C36—C46	3.9 (2)	C22—C23—C24—C25	0.3 (2)
C34—C35—C36—C31	-1.5 (2)	C27—C23—C24—C25	-179.7 (1)
C34—C35—C36—C46	177.0 (1)	C23—C24—C25—N26	-0.1 (2)
C31—C36—C46—C45	148.6 (1)	C22—C21—N26—C25	0.7 (2)
C35—C36—C46—C45	-29.8 (2)	N2—C21—N26—C25	-179.4 (1)
C31—C36—C46—C41	-34.2 (2)	C24—C25—N26—C21	-0.4 (2)
C35—C36—C46—C41	147.5 (1)	C56—C51—C52—C53	0.0
C41—C46—C45—C44	-1.1 (2)	C50—C51—C52—C53	-179.4 (5)
C36—C46—C45—C44	176.3 (1)	C51—C52—C53—C54	0.0
C46—C45—C44—C43	-0.9 (2)	C52—C53—C54—C55	0.0
C45—C44—C43—C42	1.4 (2)	C53—C54—C55—C56	0.0
C44—C43—C42—C41	0.1 (2)	C54—C55—C56—C51	0.0
C43—C42—C41—C46	-2.2 (2)	C52—C51—C56—C55	0.0
C43—C42—C41—N4	178.2 (1)	C50—C51—C56—C55	179.4 (5)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N3—H3 \cdots O1	0.88 (1)	2.22 (1)	2.677 (1)	112 (1)
N3—H3 \cdots O1 ⁱ	0.88 (1)	2.31 (1)	3.108 (1)	151 (1)
N2—H2 \cdots N4	0.87 (1)	2.19 (1)	2.644 (1)	112 (1)

Symmetry codes: (i) $-x, -y+1, -z+1$.

Fig. 1

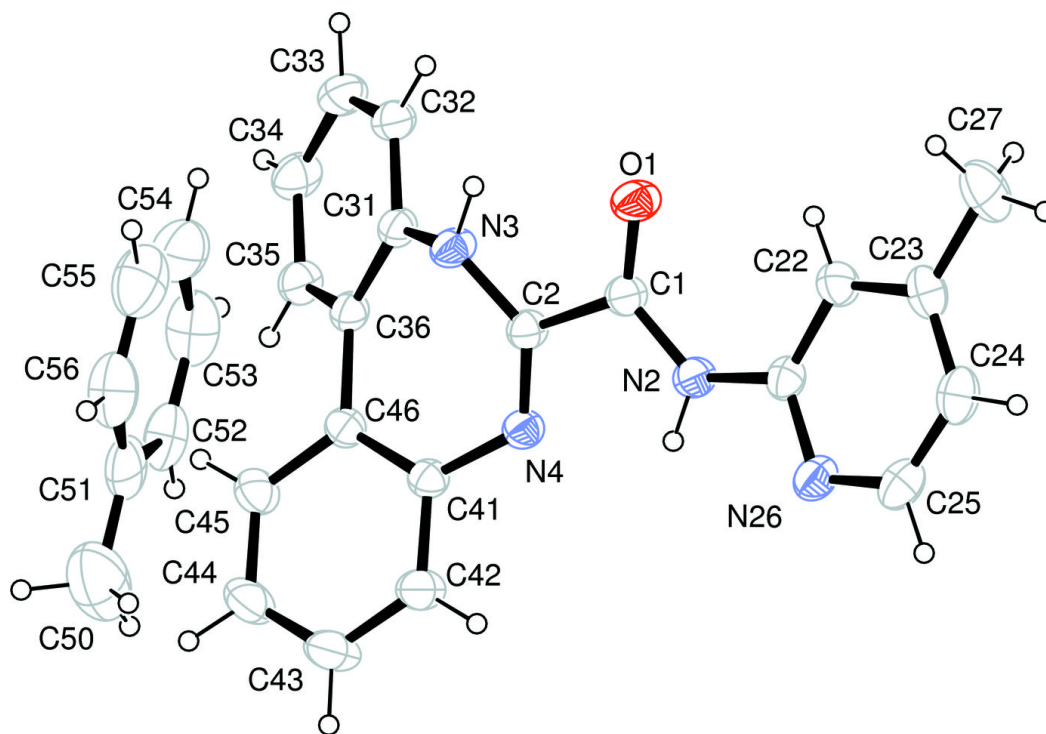


Fig. 2

